

BIOCHEMISTY

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LIPIDS



Lipids => broad group of naturally occurring molecules => include

- > fats,
- waxes,
- > sterols,
- > fat soluble vitamins,
- > monoacylglycerols,
- > diacylglycerols,
- > phospholipids and
- > others.

- compounds => generally
- > soluble in organic solvents and
- > largely **insoluble** in water.
- main biological functions of lipids :
- > energy storage,
- > structural components of cell membranes,
- > vitamins,
- **hormones** and
- > important signaling molecules.

- > Lipid is sometimes used as a synonym for **fats**.
- > fats are a subgroup of lipids called triacyl glycerols.
- Lipids also encompass molecules, such as -
- Fatty acids and their derivatives (including tri-, di- and monoacylglycerols and phospholipids), as well as other
- > sterol-containing metabolites such as cholesterol.
- Humans use various biosynthetic pathways => both degrade and synthesize lipids
- > some **essential lipids** cannot be made this way => must be obtained from the **diet**.

Lipid Classification (7 groups):

- > Fatty Acids
- > Acyl glycerols
- > Glycerophospholipids
- > Sphingolipids
- > · Sterols
- > Prenol lipids
- > Saccharo lipids

1. Fatty acids

- > consists of :
- hydrocarbon chain and
- > a terminal carboxylic acid group.
- This arrangement confers the molecule with a polar, hydrophilic end and a nonpolar, hydrophobic end => insoluble in water.
- ➤ fatty acid structure => the most fundamental categories of biological lipids => commonly used as a **building block** => more structurally complex lipids.

- > carbon chain (4 to 24 carbons) => may be saturated or unsaturated.
- > saturated fatty acid => all the carbon atoms => saturated with hydrogen atoms with general formula CH3(CH2)nCOOH where => n is an even number.
- Mono-unsaturated fatty acids => one double bond in their structure.
- > polyunsaturated fatty acids => two or more double bonds.
- ➤ double bonds in polyunsaturated fatty acids are generally separated by at least one methylene group.

- double bond => there is the possibility of => either a
 cis or trans geometric isomerism => significantly affects
 => molecule's molecular configuration.
- Cis-double bonds => cause the fatty acid chain to bend => an effect that is more pronounced when more double bonds are there in a chain.
- > plays important role => structure and function of **cell** membranes.

- Naturally occurring fatty acids => cis configuration, although trans form => natural and partially hydrogenated fats and oils.
- > Shorter the chain => lower is the melting temperature
- ➤ Unsaturated fatty acids => lower melting temperatures than saturated fatty acids of same chain length.

2. Glycerolipids

- > Glycerolipids are composed mainly of
- > mono-, di- and tri-substituted glycerols, the most well-known => triacylglycerols /triglycerides / fats.
- In these compounds => all three hydroxyl groups of glycerol => esterified => different fatty acids (Mixed Lipids).



- > They function => food store.
- > these lipids comprise => bulk of storage fat in animal tissue and oil seeds.

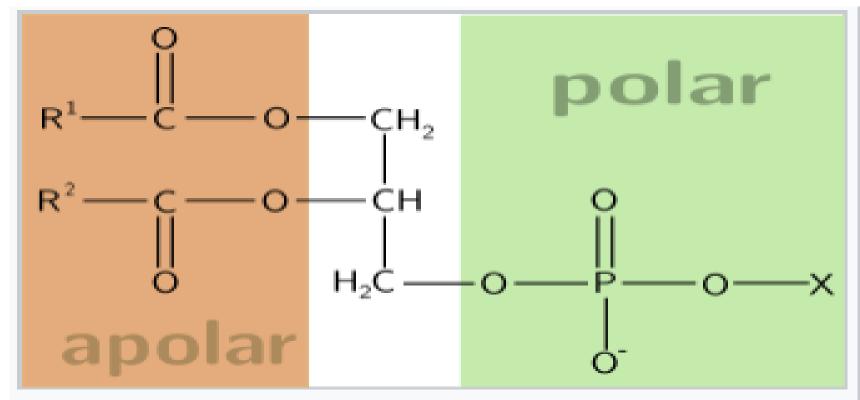
TG or fats may be => solid or liquid at room temperature => their structure and composition.

- > "Oils" => used to refer to fats => liquids at normal room temperature.
- "fats" => used to refer to fats => solids at normal room temperature.
- > "Lipids" => used to refer to => both liquid and solid fats, along with other related substances.

3. Glycerophospholipids

- > phospholipids :
- > key components of the lipid bilayer of cells
- involved in **metabolism** and
- > cell signaling.
- Neural tissue (including brain) contains relatively high amounts
 => glycerophospholipids and
- alterations in their composition has been implicated in => various neurological disorders.

- > PL found in biological membranes :
- > Phosphatidyl choline (PC or lecithin)
- > phosphatidyl ethanolamine (PE) and
- > Phosphatidyl serine (PS).
- ➤ Plasmalogens => a type of glycerolipids contain => fatty alcohol at C-1 of Sn glycerol with double bond instead of a fatty acid.



Glycerophospholipids have three components: fatty acid lipid groups (orange), glycerol (white), and phosphate ester (green)

L-Glycerol 3--phosphate

Amphiphilic

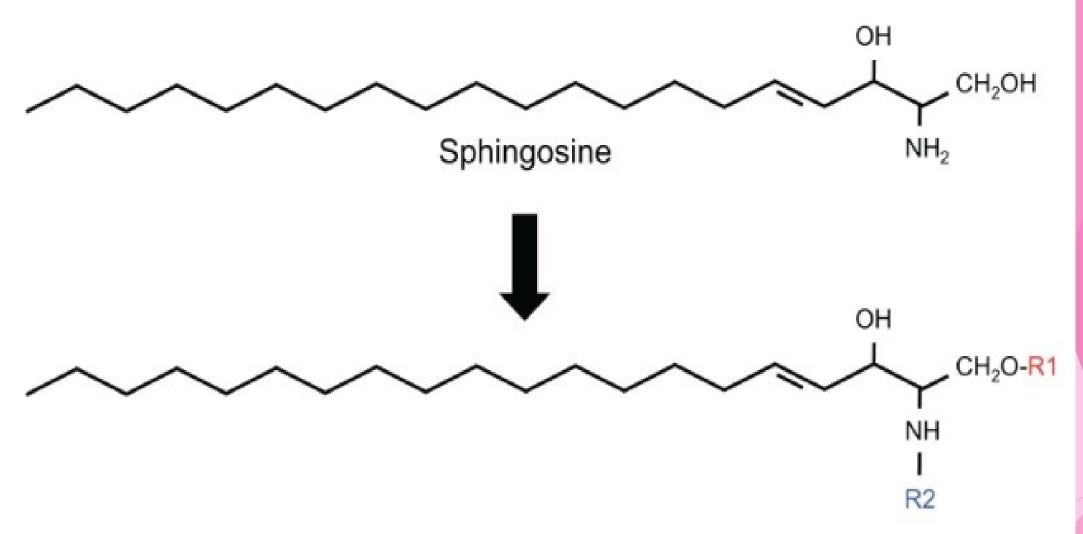
Name of glycoroglospholipid	Name of X	Ferman la vil'N	Pet champs (as p40 f)
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Phosphartidylethus elamino	little an elametro	— сп ₂ —сп ₂ —8п ₄	0
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Cardidipin	Phosphacidyli- gly-sorol	CHOR O	-6
		314-0-B-10	

Glycerophospholipids

4. Sphingolipids

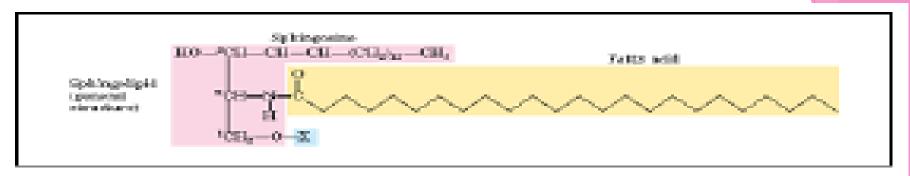
- complex family of compounds that share => common structural feature => a sphingoid base backbone (synthesized de novo from the amino acid serine and a long- chain fatty acyl CoA) => then converted into =>
- > ceramides,
- > phosphosphingolipids,
- glycosphingolipids and
- > other compounds.

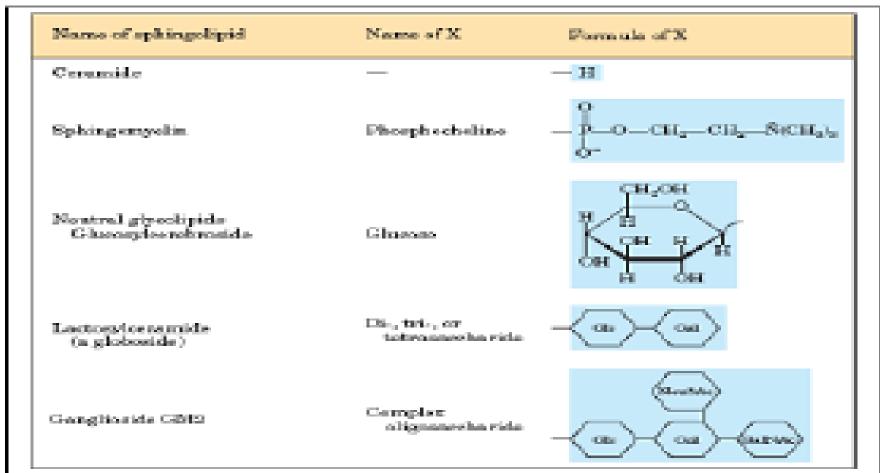
- > major sphingoid base of mammals is commonly referred to as => sphingosine.
- Ceramides (N-acylsphingoid bases) are a major subclass of sphingoid base derivatives with an amide-linked fatty acid.
- The fatty acids are typically saturated or monounsaturated with chain lengths from 16 to 26 carbon atoms.



Sphingolipid

Structure of sphingolipids. In sphingolipids, the hydrophobic region consists of a longchain sphingoid base with generally 18 carbons, such as sphingosine, which is linked to the acyl group of a fatty acid via an amide bond (R2). The hydrophilic region (R1) consists in the simplest case of a hydroxyl group in the case of ceramide.





Sphingolipids

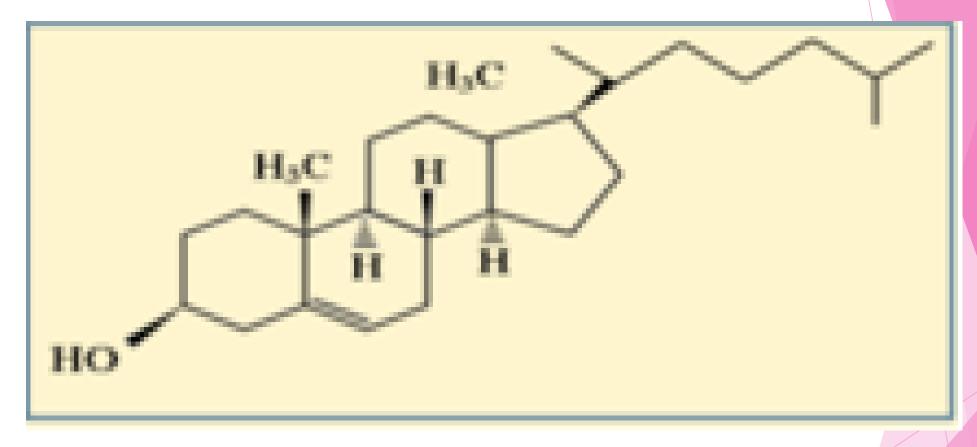
- > Functions of mammalian sphingolipids
- > Sphingolipids => protect the cell surface against harmful environmental factors by forming a mechanically stable and chemically resistant outer leaflet of the <u>plasma membrane lipid bilayer</u>.
- Certain complex <u>glycosphingolipids</u> => involved in specific functions => <u>cell recognition and signaling</u>.
- Cell recognition depends => on physical properties of sphingolipids
- signaling involves specific interactions of the glycan structures of glycosphingolipids with similar lipids present on neighboring cells or with proteins.

5. Sterols

- > Sterol lipids => cholesterol and its derivatives => important component of membrane lipids, along with the glycerophospholipids and sphingomyelins.
- > Steroids => derived from the same fused four-ring core structure => have different biological roles as hormones and signaling molecules.
- eighteen-carbon(C18) steroids => include the estrogen family
 C19 steroids => comprise the androgens such as testosterone and androsterone.

- > C21 subclass => includes the progestogens as well as glucocorticoids and mineralocorticoids.
- secosteroids => comprising various forms of vitamin
 D => are characterized by cleavage of the B ring of the core structure.
- ➤ Other examples of sterols => bile salt and their conjugates, which in mammals are => oxidized derivatives of cholesterol and are synthesized in the liver.

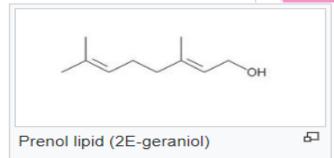
cholesterol



Sterols are <u>steroids</u> in which one of the hydrogen atoms is substituted with a <u>hydroxyl group</u> in the carbon chain. They have in common with steroids the same fused four-ring core structure.

6. Prenol lipids

- > Prenol (3-methyl-2-buten-1-ol) => **natural alcohol**.
- > one of the simplest terpenes.



- Prenol lipids => synthesized from => 5-carbon precursors isopentenyl diphosphate and dimethylallyl diphosphate => produced mainly via => mevalonic acid (MVA) pathway.
- > simple **isoprenoids** (linear alcohols, diphosphates) are formed by successive addition of C5 units and => are classified according to number of these terpene units.

- > Structures containing > than 40 carbons are known as polyterpenes.
- ➤ Carotenoids are important simple isoprenoids => function as antioxidants and as precursors of vitamin A.
- Another biologically important class of molecules is exemplified by the quinones and hydroquinones => contain an isoprenoid tail attached to a quinonoid core of non-isoprenoid origin.
- ➤ Vitamin E, vitamin K and ubiquinones => examples of this class.

7. Saccharolipids

- compounds in which fatty acids are directly linked to sugar backbone => forming structures => compatible with membrane bilayers.
- ➤ In saccharolipids => a monosaccharide substitutes for the glycerobackbone present in glycerolipids and glycerophospholipids.
- most familiar saccharolipids => acylated glucosamine precursors of the Lipid A component of the lipopolysaccharides =>
- Gram-negative bacteria

Common Fatty Acids

Chemical Names and Descriptions of some Common Fatty Acids					
Common Name	Carbon Atoms		Scientific Name	Sources	
Butyricacid	4	0	butanoic acid	butterfat	
Caproid Add	6	0	hexanoic acid	butterfat	
Caprylio Acid	8	0	octanoic acid	coconut oil	
Capric Acid	1.0	0	decanoic acid	coconut oil	
Lauric Acid	1.2	0	dodecanoic acid	coconut oil	
Myristic Acid	14	0	tetrade can oic acid	palm kernel oil	
Palmitic Acid	1.6	0	hexade can oic acid	palm oil	
Palmitoleic Acid	1.6	1	9-hexadecenoic acid	animal fats	
Stearic Acid	1 B	0	octadecano ic acid	animal fats	
Oleic Acid	1 B	1	9-octadecencie acid	olive oil	
Vaccenic Acid	1.6	1	11-octadecenoicacid	butterfat	
Linoleic Acid	1 B	2	9,12-octad ecadienoic acid	grape seed oil	
Alph a-Linolenic Acid (ALA)	1 B	3	9,12,15-octadecatrienoic acid	flaxseed (linseed) oil	
Gamma-Linolenic Acid (GLA)	18	3	6,9,12-octadecatrienoic acid	borage oil	
Arachidic Acid	2 D	0	eicasanoic acid	peanut oil, fish oil	
Arachidonic Acid (AA)	20	4	5,8,11,14-eicosatetraenoidacid	liver fats	
ЕРА	2 0	5	5,8,11,14,17-eicosapentaenoid acid	fish oil	

THANKS